

This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

GENERAL APPROACH TO THE SYNTHESIS OF SYMMETRICAL 1,3,2-DIAZAPHOSPHOCANES

Alexandra I. Zavalishina^a; Elena I. Orzhekovskaya^a; Natalja M. Selezneva^a; Larisa K. Vasyanina^a; Vitaly K. Belsky^b; Eduard E. Nifantsev^a

^a Faculty of Chemistry# knin Pedagogical State University, Moscow, RUSSIA ^b Karpov Research Institute of Physical Chemistry, Moscow, RUSSIA

To cite this Article Zavalishina, Alexandra I. , Orzhekovskaya, Elena I. , Selezneva, Natalja M. , Vasyanina, Larisa K. , Belsky, Vitaly K. and Nifantsev, Eduard E.(2000) 'GENERAL APPROACH TO THE SYNTHESIS OF SYMMETRICAL 1,3,2-DIAZAPHOSPHOCANES', Phosphorus, Sulfur, and Silicon and the Related Elements, 161: 1, 205 — 211

To link to this Article: DOI: 10.1080/10426500008042108

URL: <http://dx.doi.org/10.1080/10426500008042108>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

GENERAL APPROACH TO THE SYNTHESIS OF SYMMETRICAL 1,3,2-DIAZAPHOSPHOCANES

ALEXANDRA I. ZAVALISHINA^a, ELENA I. ORZHEKOVSKAYA^a,
NATALJA M. SELEZNEVA^a, LARISA K. VASYANINA^a,
VITALY K. BELSKY^b and EDUARD E. NIFANTYEV^{a*}

^a*Faculty of Chemistry, Lenin Pedagogical State University, Nesvizhskii per. 3,
Moscow, 119021 RUSSIA and* ^b*Karpov Research Institute of Physical Chemistry,
ul. Vorontsovo pole 10, Moscow, 103064 RUSSIA*

(Received July 7, 1999; Revised October 10, 1999)

Synthesis of eight-membered phosphorus-nitrogen heterocycles has been performed by phosphocyclization of the corresponding diamines using different derivatives of phosphorous acid. The structures of diazaphosphocanes were supported by elemental analysis; ¹H, ¹³C, and ³¹P NMR spectroscopy; and x-ray diffraction analysis.

Keywords: Diazaphosphocan

1,3,2-Diazaphosphocyclanes containing 4–7-membered phosphorus-nitrogen heterocycles have been studied in detail^{1–3}. These compounds are shown to have peculiar chemical features and to be promising for development of fine organic synthesis.

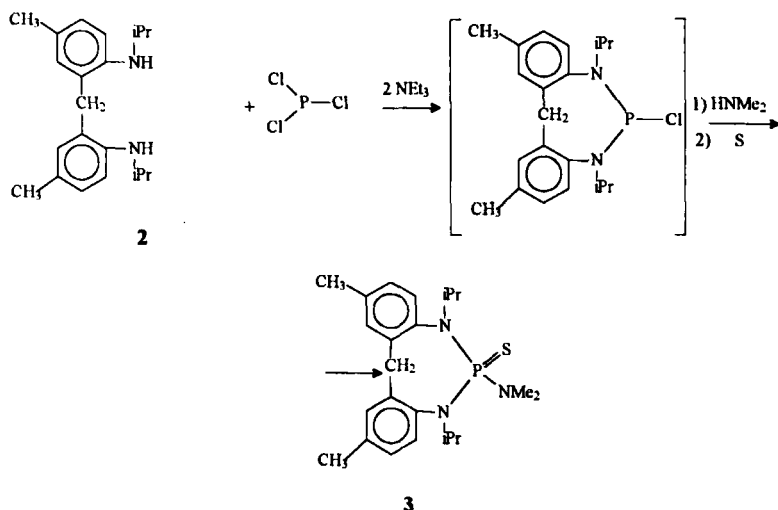
The reaction of *N,N*-dimethyl-*p*-toluidine with phosphorus oxychloride and phosphorus thiochloride resulting in formation of eight-membered phosphorus-nitrogen heterocycles was studied previously by Shaw *et al.*^{4–6}. No further publications concerned with 1,3,2-diazaphosphocanes are available.

This is to give notice that 1,3,2-diazaphosphocanes belonging to different chemical types can be obtained by the direct cyclophosphorylation of 2,2'-methylene-bis-*p*-toluidine (1) and its derivatives with phosphorus

* Correspondence Author.

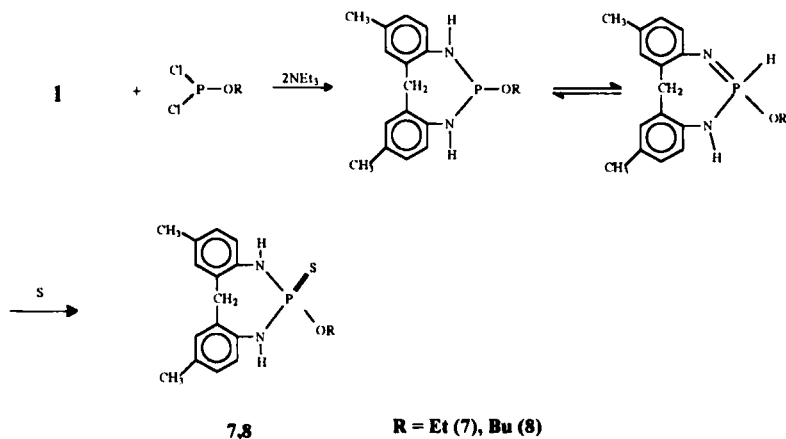
acid chlorides. The high lability of 1,3,2-diazaphospho(III)canes should be noted, in distinction from the 5–7 membered analogues known to date, which significantly complicated the operations. So, the reaction of *N,N'*-diisopropyl-2,2'-methylene-bis-*p*-toluidine (**2**) with phosphorus trichloride in the presence of triethylamine gives 2-chloro-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocane. This is very unstable compound. It was stabilized using dimethylaminolysis in combination with sulfurization. Resulting 2-dimethylamido-2-thio-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocane (**3**) was isolated with a yield of 45%

The structure of this compound was supported by ^1H and ^{31}P NMR spectroscopy, and its spatial organization was confirmed by x-ray diffraction analysis (see figure).



Diamine **2** was also subjected to cyclophosphorylation with dichlorides of ethyl-, phenyl-, and diethylamidophosphorous acids. The corresponding 2-substituted 1,3,2-diazaphosphocanes were obtained and isolated as thiophosphates (**4–6**).

The phosphocyclization of diamine **1** with alkylidichlorophosphites follows an unusual pathway. Mixtures of two tautomeric forms are the products:



This equilibrium is evidenced by the appearance of two signals at 113 and 6.23 ($J_{\text{PH}} 614 \text{ Hz}$) in the ^{31}P NMR spectrum. It is important that the only product (**7, 8**) forms at the treatment of the system obtained by sulfur in both cases.

The obtained eight-membered cyclic systems containing a trivalent phosphorus atom are very labile. For example, the interaction between the primary diamine **1** and phenyldichlorophosphite resulted in formation of the corresponding 2-phenoxy derivative, for which the above prototropy was not observed. However, the compound obtained was readily hydrolyzed with formation of an incomplete diazaphosphocan amide (**9**), when chromatographed on silica gel.

Some of the compounds studied are susceptible to dimerization, as was shown during their study using mass spectrometry.

$\text{P-N}(1) = 1.648(4)$, $\text{P-N}(3) = 1.663(3)$, $\text{P-N}(2) = 1.667(3)$, $\text{N}(2)\text{-C}(1) = 1.442(5)$, $\text{N}(2)\text{-C}(16) = 1.501(5)$, $\text{N}(3)\text{-C}(9) = 1.439(2)$, $\text{N}(3)\text{-C}(19) = 1.500(5)$, $\text{C}(6)\text{-C}(7) = 1.507$, $\text{C}(7)\text{-C}(8) = 1.512(5)$, $\text{C}(1)\text{-C}(6) = 1.394$, $\text{C}(8)\text{-C}(9) = 1.400(5)$.

$\text{N}(1)\text{-P-N}(3) = 110.37(17)$, $\text{N}(1)\text{-P-N}(2) = 102.95(17)$, $\text{N}(3)\text{-P-N}(2) = 103.92(16)$, $\text{N}(1)\text{-P-S} = 111.15(15)$, $\text{N}(3)\text{-P-S} = 110.97(12)$, $\text{N}(2)\text{-P-S} = 116.98(12)$, $\text{C}(22)\text{-N}(1)\text{-P} = 125.6(4)$, $\text{C}(23)\text{-N}(1)\text{-P} = 120.8(4)$, $\text{C}(1)\text{-N}(2)\text{-C}(16) = 118.0(3)$, $\text{C}(1)\text{-N}(2)\text{-P} = 122.1(2)$, $\text{C}(16)\text{-N}(2)\text{-P} = 117.8(3)$, $\text{C}(9)\text{-N}(3)\text{-C}(19) = 118.1(3)$, $\text{C}(9)\text{-N}(3)\text{-P} = 121.9(2)$, $\text{C}(19)\text{-N}(3)\text{-P}$

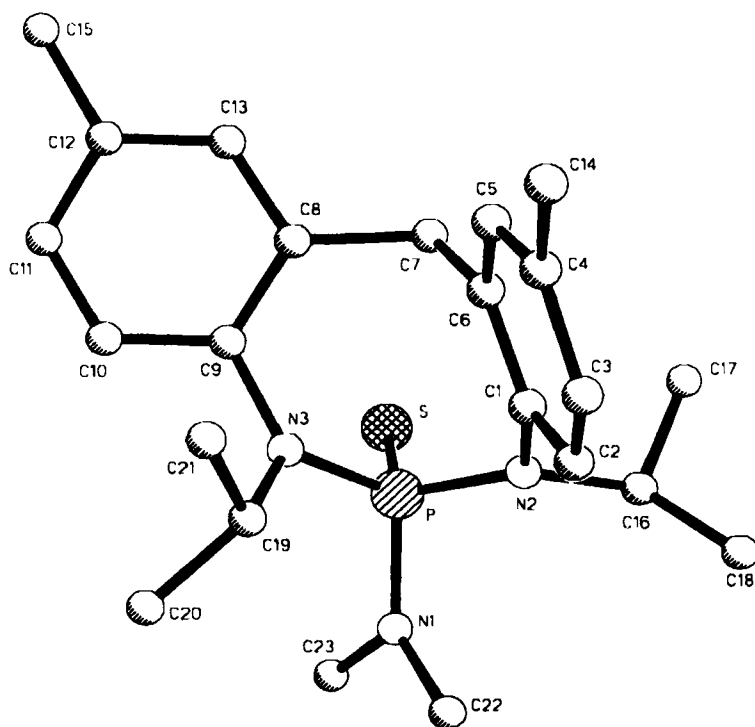


FIGURE The structure of **3** from the x-ray diffraction analysis data

= 119.6(3), C(2)-C(1)-N(2) = 119.3(3), C(6)-C(1)-N(2) = 121.8(3), C(1)-C(6)-C(7) = 122.6(4), C(6)-C(7)-C(8) = 114.2(3), C(8)-C(9)-N(3) = 122.0(3)

EXPERIMENTAL

^{31}P NMR spectra were recorded on a Bruker WP-80 instrument; ^1H NMR spectra were recorded on a Bruker WM-250 spectrometer. Mass-spectrometric studies were performed by the electron impact method on a Kratos MS-890 instrument. X-ray diffraction analysis was carried out on a Syntex P-1. An X-ray analysis of **3** was performed on a Syntex P-1 diffractometer using $\text{Cu-K}\alpha$ radiation ($\lambda = 1.54178 \text{ \AA}$). Crystals are orthorhombic, space

group *Pbca*, $C_{23}H_{34}N_3PS$, Z 8, d_{calc} 1.179 g/cm³, a 14.908(3), b 16.547(3), c 18.988(3) Å. The structure was solved by a direct method and was refined anisotropically basing on 2148 independent reflections with $I \geq 2\sigma(I)$ to a final $R=0.029$. All calculations have been done using SHELX97 package.

General procedure

Phosphorus trichloride or dichloride of alkyl- (aryl- or dialkylamido-) phosphorous acid was added slowly to a solution of 0.01 mol of diamine (**1**, **2**) and 0.02 mol of triethylamine in dry benzene under stirring at 5–10°C. The reaction mixture was stirred at the room temperature; triethylamine hydrochloride was filtered off, and the raw cyclic reaction product was used. In order to obtain the corresponding thioderivatives (**3–8**), sulfur was added. The reaction mixture was stirred for 2–3 h; the solvent was removed, and the target product was isolated either by recrystallization from benzene (**4**) or using the column chromatography on silica gel 45/75 (**3**, **5–8**).

2-Thio-2-dimethylamido-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (3)

Yield = 45%, m.p. 141–142°C. $\delta_P(CHCl_3)$ 69.80. MS, m/e : 415. ¹H NMR ($CDCl_3$, δ): 1.24 d (6H, CH_3^a , $^3J_{HH}$ 6.4 Hz), 1.3 d (6H, CH_3^B , $^3J_{HH}$ 7.83 Hz), 2.33 d (6H, $N(CH_3)_2$, $^3J_{HP}$ 10.9 Hz), 2.23 s (3H, CH_3^{11}), 2.27 s (3H, $CH_3^{11'}$), 3.09 m (1H, CH^{iPr} , $^3J_{HP}$ 15.3 Hz, $^3J_{HH}$ 7.83 Hz, $^3J_{HH}$ 6.4 Hz), 3.74 d (1H, CH_2 , $^2J_{HH}$ 14.74 Hz), 4.06 m (1H, CH^{i-Pr} , $^3J_{HP}$ 15.3 Hz, $^3J_{HH}$ 7.83 Hz, $^3J_{HH}$ 6.4 Hz), 4.7 d (1H, CH_2 , $^2J_{HH}$ 14.74 Hz), 6.56–7.09 m (6H, Ar). Found, %: C 66.3; H 8.39; P 7.07 $C_{23}H_{34}N_3PS$. Calcd., %: C 66.5; H 8.19; P 7.14.

2-Thio-2-ethoxy-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (4)

Yield = 45%, m.p. 164–166°C. $\delta_P(CHCl_3)$ 71.51. MS, m/e : 416. ¹H NMR ($CDCl_3$, δ): 0.25 (0.28) dd (6H, $CH_3^{a,B}$, $^3J_{HH}$ 2.56 Hz), 1.1 t (3H, CH_3CH_2O , $^3J_{HH}$ 6.83 Hz), 1.35 d (3H, CH_3^a , $^3J_{HH}$ 6.83 Hz), 1.44 d (3H, CH_3^B , $^3J_{HH}$ 6.83 Hz), 2.28 s (6H, $CH_3^{11,11'}$), 3.77 m (1H, CH^{i-Pr} , $^3J_{HP}$ 15.37 Hz, $^3J_{HH}$ 2.56 Hz, $^3J_{HH}$ 6.83 Hz), 3.90 m (1H, CH^{i-Pr} , $^3J_{HP}$ 15.37 Hz,

$^3J_{\text{HH}}$ 2.56 Hz, $^3J_{\text{HH}}$ 6.83 Hz), 3.95 m (2H $\text{CH}_3\text{CH}_2\text{O}$, $^3J_{\text{HP}}$ 8.53 Hz, $^3J_{\text{HH}}$ 6.83 Hz), 4.04 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.51 Hz), 4.17 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.51 Hz), 6.94–7.11 m (6H, Ar). Found, %: C 65.90; H 8.26; P 6.26. $\text{C}_{23}\text{H}_{33}\text{N}_2\text{OPS}$. Calcd., %: C 66.3; H 7.9; P 7.4.

2-Thio-2-phenoxy-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (5)

Yield = 30%, m.p. 165–166°C. $\delta_{\text{p}}(\text{C}_6\text{H}_6)$ 67.57. ^1H NMR (C_6H_6 , δ): 0.36 (0.40) dd (6H, $\text{CH}_3^{\text{a,b}}$, $^3J_{\text{HH}}$ 6.8 Hz), 1.39 d. (3H, CH_3^{a} , $^3J_{\text{HH}}$ 6.4 Hz), 1.47 d. (3H, CH_3^{b} , $^3J_{\text{HH}}$ 6.4 Hz), 2.06 s (3H, CH_3^{11}), 2.1 s (3H, $\text{CH}_3^{11'}$), 4.1 m (1H, CH^{iPr} , $^3J_{\text{HP}}$ 15 Hz, $^3J_{\text{HH}}$ 6.8 Hz, $^3J_{\text{HH}}$ 6.4 Hz), 4.20 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.52 Hz), 4.28 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.52 Hz), 4.64 m (1H, CH^{iPr} , $^3J_{\text{HP}}$ 15 Hz, $^3J_{\text{HH}}$ 6.8 Hz, $^3J_{\text{HH}}$ 6.4 Hz), 6.78–7.34 m (11H, Ar).

2-Thio-2-diethylamido-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (6)

Yield = 42%, m.p. 170–171°C. $\delta_{\text{p}}(\text{CHCl}_3)$ 70.97. MS, m/e: 443. ^1H NMR (CDCl_3 , δ): 1.23 d (12H, CH_3^{iPr} , $^3J_{\text{HH}}$ 7.46 Hz), 1.27 m (6H, CH_3CH_2 , $^3J_{\text{HH}}$ 6.83 Hz), 2.28 s (3H, CH_3^{11}), 2.32 s (3H, $\text{CH}_3^{11'}$), 3.16 q (1H, CH^{iPr} , $^3J_{\text{HH}}$ 7.46 Hz), 3.25 m (2H, $-\text{CH}_2-\text{CH}_3$, $^3J_{\text{HH}}$ 6.83 Hz, $^3J_{\text{HP}}$ 12.81 Hz), 3.48 m (2H, $-\text{CH}_2-\text{CH}_3$, $^3J_{\text{HH}}$ 6.83 Hz, $^3J_{\text{PH}}$ 13.23 Hz), 3.79 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.72 Hz), 4.03 q (1H, CH^{iPr} , $^3J_{\text{HH}}$ 6.83 Hz), 4.64 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.72 Hz), 6.97–7.12 m (6H, Ar).

2-H-2-Oxo-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (9)

Yield = 35%. m.p. 172–173°C. $\delta_{\text{p}}(\text{CHCl}_3)$ 10.31. MS, m/e: 272. ^1H NMR (CDCl_3 , δ): 2.29 s (6H, $\text{CH}_3^{11,11'}$), 3.93 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.29 Hz), 4.03 d (1H, CH_2 , $^2J_{\text{HH}}$ 14.29 Hz), 4.97 s (2H, $\text{NH}^{1,3}$, $^3J_{\text{HH}}$ 2.2 Hz), 7.68 d (1H, J_{PH} 614 Hz), 6.86–7.11 m (6H, Ar).

Note: Letters *a* and *e* denote two magnetically nonequivalent methyl groups in isopropyl radicals.

References

1. X.-B. Ma, J.Z. Zhang, *Phosphorus, Sulfur, Silicon*, **53**, 233 (1990).
2. M.A. Pudovik, V.V. Ovchinnikov, R.A. Cherkasov, and A.N. Pudovik, *Usp. Khim.*, **52**, 640 (1983).
3. E.E. Nifantiev, A.I. Zavalishina, E.I. Orzhkovskaya, N.N. Nurkulov, L.K. Vasjanina, A.R. Bekker, V.K. Belsky, and A.I. Stash, *Phosphorus, Sulfur, Silicon*, **123**, 89 (1997).

4. C.Y. Cheng, R.A. Shaw, T.S. Cameron, and C.K. Prout, *J. Chem. Soc., Chem. Commun.*, 616 (1968).
5. C.Y. Cheng and R.A. Shaw, *Phosphorus, Sulfur, Silicon*, **26**, 185 (1986).
6. T. Demir and R.A. Shaw, *J. Chem. Soc. Perkin Trans.*, **1**, 1547 (1987).